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see "The Free-Response Sections ? Student Presentation" in the AP Physics Course Description. 4. The scoring guidelines typically show numerical results using the value  $g = 9.8 \text{ m/s}^2$ , but use of  $10 \text{ m/s}^2$  is of course also acceptable. Solutions usually show numerical answers using both values when they are significantly different. 5.

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25 10. Mgd M M Mqu u =+ = u. 2. uq == (10 7  
sin 10 7 9.8 m s 4.0 m sin30) gd ( ) (2) ( ) ( )  
(?) For a correct numerical answer . 1 point .  
u = 5.3 m s. Alternate solution (kinematics  
method) Alternate points . For determination

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of the linear acceleration in terms of mass and frictional force : 1 point . aFM = 52 f from work shown in part (b)

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10 Circulation, but could anyone help me out with the questions. I just really want to make sure that my answers are correct. Here are the questions! Questions Part A and B 1. Q. Explain why blood pressure and heart rate differ when measured in a reclining position and in a standing position. A 2.

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Question 3 (9 points)  $8 \text{H}^+(\text{aq}) + 4 \text{Cl}^-(\text{aq}) + \text{MnO}_4^-(\text{aq}) \rightarrow 2 \text{Cl}_2(\text{g}) + \text{Mn}^{2+}(\text{aq}) + 4 \text{H}_2\text{O}(\text{l})$   $\text{Cl}_2(\text{g})$  can be generated in the laboratory by reacting potassium permanganate with an acidified solution of sodium chloride. The net-ionic equation for the reaction is given above. (a) A 25.00 mL sample of 0.250 M NaCl reacts completely with excess  $\text{KMnO}_4$

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## AP CHEMISTRY 2010 SCORING GUIDELINES

Score: 10. The response earned 1 point in part (a) for correctly plotting the means on a bar graph. The response earned 1 point in part (a) for appropriate labels, units, and scaling. The response earned 1 point in part (a) for correctly plotting the error bars.

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## AP Biology 2017 FRQ 1 Student Samples

Briefly explain your answer. I An unknown solid has a melting point of 112–114°C. Based on data from Table 1 above and the mixed melting point technique of this experiment, you can narrow down the possibilities to three compounds. List the compounds b) A mixed melting point of the unknown with acetanilide gives a melting point of 113–114°C.

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Melting Point Experiment Experiment #10 -  
Melting ...

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1. To observe the reaction to be studied, transfer 10-mL of 1.5% (0.44 M) H<sub>2</sub>O<sub>2</sub> into a 50-mL glass beaker and add 1 mL of the freshly made catalase solution. The bubbles coming from the reaction mixture are O<sub>2</sub>, which results from the breakdown of H<sub>2</sub>O<sub>2</sub> by catalase. Be sure to keep the freshly made catalase solution on ice at all times.

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AP BIOLOGY : Enzyme Catalysis Lab

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discuss a topic is asking you for something broader than a mere description. A discussion is more like a conversation about ideas, and ...

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AP CHEMISTRY BY SATELLITE 9 - 7 INSTRUCTOR LAB MANUAL EXPERIMENT 9: SOLUBILITIES Pre-lab

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Questions: Answer these questions before coming to class. They introduce you to several important ideas that you will use in this experiment. You must turn-in this exercise before you will be allowed to begin the experiment.

The book itself contains chapter-length subject reviews on every subject tested on the AP Chemistry exam, as well as both sample multiple-choice and free-response questions at each chapter's end. Two full-length practice tests with detailed answer explanations are included in the book.

Test prep for the AP Chemistry exam, with 100% brand-new content that reflects recent exam changes Addressing the major overhaul that the College Board recently made to the AP Chemistry exam, this AP Chemistry test-prep guide includes completely brand-new content tailored to the exam, administered every May. Features of the guide include review sections of the six "big ideas" that the new exam focuses on: Fundamental building blocks Molecules and interactions Chemical reactions Reaction rates Thermodynamics Chemical equilibrium Every section includes review questions and answers. Also included in the guide are two full-length practice tests as well as a math review section and sixteen discrete laboratory exercises to

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prepare AP Chemistry students for the required laboratory experiments section on the exam.

The process of investigation is dynamic and fluid. The approach must be reasonable and the investigator flexible. However, in order to be successful, every investigation must have a meaningful purpose and be executed ethically and lawfully. Inevitably, employers must invest time, money, and patience to ensure they obtain demonstrable and actionable

Workers' Compensation Index: A Topical Guide to California Workers' Compensation Law (formerly Work Comp Index) is a valuable research tool for anyone involved in handling California workers' compensation cases. The book and its Table of Contents, both of which are set up alphabetically by subject, serve as a quick reference tool and a starting point for more in-depth legal research. The California workers' compensation system has changed so rapidly in substantive and procedural areas that its participants have been inundated with information from many sources. This topical guide leads its users simply, directly, and quickly to the applicable statutory and case law and rules and to a wide variety of key publications, texts, and treatises that have treated the entire range of issues and developments in the field.

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Provides techniques for achieving high scores on the AP biology exam and includes two full-length practice tests.

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