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below was used to  
prepare dry  
hydrogen chloride  
gas, and investigate  
its effect on heated

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Considering the half-  
equations, and  
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Question 30, which  
related to an  
experimental  
context less

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PROCEDURE: Rinse  
and fill the burette  
with the solution

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C-10 (KMnO<sub>4</sub>).

Pipette out 20 ml or  
25 ml of C-11

(hydrated  
ammonium iron(II)  
sulphate) into a  
clean conical flask.  
To this, add 20ml of  
dilute sulphuric acid  
C-12, specially  
provided for  
titration.

CHEMISTRY

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Structure of Matter



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few marks in Paper

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Reaction (about 2

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